

## Limiting Reactant Worksheet #2

1. Given the following reaction: (Balance the equation first!)



- If you start with 14.8 g of  $\text{C}_3\text{H}_8$  and 3.44 g of  $\text{O}_2$ , determine the limiting reactant.
- Determine the number of moles of carbon dioxide produced.
- Determine the number of grams of  $\text{H}_2\text{O}$  produced.
- Determine the number of grams of excess reactant left.

2. Given the following equation:



- If 10.0 g of  $\text{Al}_2(\text{SO}_4)_3$  is reacted with 10.0 g of  $\text{NaOH}$ , determine the limiting reactant.
- Determine the number of moles of  $\text{Al}(\text{OH})_3$  produced.
- Determine the number of grams of  $\text{Na}_2\text{SO}_4$  produced.
- Determine the number of grams of excess reactant left over in the reaction.

3. Given the following equation:



- If 25.4 g of  $\text{Al}_2\text{O}_3$  is reacted with 10.2 g of  $\text{Fe}$ , determine the limiting reactant.
- Determine the number of moles of  $\text{Al}$  produced.
- Determine the number of grams of  $\text{Fe}_3\text{O}_4$  produced.
- Determine the number of grams of excess reactant left over in the reaction.

4. Consider the reaction



- 80.0 grams of  $\text{I}_2\text{O}_5$ , reacts with 28.0 grams of carbon monoxide,  $\text{CO}$ . Determine the mass of iodine  $\text{I}_2$ , which could be produced?
- If, in the above situation, only 0.160 moles, of iodine,  $\text{I}_2$  was produced.
  - What mass of iodine was produced?
  - What percentage yield of iodine was produced?

5. Zinc and sulfur react to form zinc sulfide according to the equation



If 25.0 g of zinc and 30.0 g of sulfur are mixed,

- Which chemical is the limiting reactant?
- How many grams of  $\text{ZnS}$  will be formed?
- How many grams of the excess reactant will remain after the reaction is over?

6. Which element is in excess when 3.00 grams of  $\text{Mg}$  is ignited in 2.20 grams of pure oxygen? What mass is in excess? What mass of  $\text{MgO}$  is formed?

7. How many grams of  $\text{Al}_2\text{S}_3$  are formed when 5.00 grams of  $\text{Al}$  is heated with 10.0 grams  $\text{S}$ ?

# Limiting Reagent Worksheet 202 Answer Key

**Robert R. Redfield**



**Limiting Reagent Worksheet 202 Answer Key:**

## Unveiling the Magic of Words: A Report on "**Limiting Reagent Worksheet 202 Answer Key**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Limiting Reagent Worksheet 202 Answer Key**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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