

# Limiting & Excess Reactants

## Limiting & Excess

### Limiting Reactant

- The reactant that will be used completely
- Once there is no more of this reactant, the reaction will stop
- It limits the amount of product that can be produced

### Excess Reactant

- The reactant that will not be used completely
- You will have extra of this reactant
- The reaction could keep going if you had more of the other reactant



## Finding Limiting/Excess

- You will have to set up two T-charts
- You'll be given two values to start with
- Determine which value will produce more of the product

# Limiting And Excess Reactants Packet Model 1

**Lauren Gardner**



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