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# Limiting Reactant Worksheet

- 1 Balance the following reaction:



- a) Suppose 100 grams of sodium chloride reacts with 200 g of barium phosphate. What is the limiting reactant?

- b) How many grams of excess reactant are left?

- 2 Acrylonitrile ( $\text{C}_3\text{H}_3\text{N}$ ) is the starting material for the production of a kind of synthetic fiber acrylic. It can be made from propylene ( $\text{C}_3\text{H}_6$ ) by reacting with nitric oxide (NO) as follows:



What mass of  $\text{C}_3\text{H}_3\text{N}$  can be made when 21.6 g of propylene react with 21.6 g of nitric oxide?

# Limiting Reactant Work Sheet

**Lei Shi**



## Limiting Reactant Work Sheet:

*General Chemistry Workbook* Daniel C. Tofan, 2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

**ChemDiscovery Teacher Edition** Olga I. Agapova, 2002

**Holt Chemistry** Holt Rinehart & Winston, 2003-01-24

*Working with Chemistry* Donald J. Wink, Sharon Fetzer-Gislason, Julie Ellefson Kuehn, 2004-02-20 With this modular laboratory program students build skills using important chemical concepts and techniques to the point where they are able to design a solution to a scenario drawn from a professional environment The scenarios are drawn from the lives of people who work with chemistry every day ranging from field ecologists to chemical engineers and include many health professionals as well

*The Thermodynamics of Phase and Reaction Equilibria* Ismail Tosun, 2021-06-17 The Thermodynamics of Phase and Reaction Equilibria Second Edition provides a sound foundation for understanding abstract concepts of phase and reaction equilibria e g partial molar Gibbs energy fugacity and activity and shows how to apply these concepts to solve practical problems using numerous clear examples Available computational software has made it possible for students to tackle realistic and challenging problems from industry The second edition incorporates phase equilibrium problems dealing with nonideal mixtures containing more than two components and chemical reaction equilibrium problems involving multiple reactions Computations are carried out with the help of Mathcad Clear layout coherent and logical organization of the content and presentation suitable for self study Provides analytical equations in dimensionless form for the calculation of changes in internal energy enthalpy and entropy as well as departure functions and fugacity coefficients All chapters have been updated primarily through new examples Includes many well organized problems with answers which are extensions of the examples enabling conceptual understanding for quantitative real problem solving Provides Mathcad worksheets and subroutines Includes a new chapter linking thermodynamics with reaction engineering A complete Instructor's Solutions Manual is available as a textbook resource

*Handbook on Material and Energy Balance Calculations in Material Processing, Includes CD-ROM* Arthur E. Morris, H. Alan Fine, Gordon Harold Geiger, 2011-09-06 This book approaches the subject of material and energy balances from two directions First it emphasizes the fundamental principles of the conservation of mass and energy and the consequences of these two principles Second it applies the techniques of computational chemistry to materials processing and introduces new software developed by the author especially for material and heat balances The third edition reflects the changes in the professional engineer's practice in the last 30 years reflecting the dramatic shift away from metallurgical engineering and the extractive industry towards materials engineering

A large and growing number of recent graduates are employed in such fields as semiconductor processing environmental engineering and the production and processing of advanced and exotic materials for aerospace electronic and structural applications The advance in computing power and software for the desktop computer has significantly changed the way engineers make computations and the biggest change comes from the computational approach used to solve problems The spreadsheet program Excel is used extensively throughout the text as the main computational engine for solving material and energy balance equations and for statistical analysis of data The use of Excel and the introduction of the add in programs enables the study of a range of variables on critical process parameters and emphasis is placed on multi device flowsheets with recycle bypass and purge streams whose material and heat balance equations were previously too complicated to solve by the normally used hand calculator The Excel based program FlowBal helps the user set up material and heat balance equations for processes with multiple streams and units

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**A Concrete Stoichiometry Unit for High School Chemistry** Jennifer Louise Pakkala,2006 **The Educational Value of Chemical Demonstrations in the College Prep Chemistry Classroom** Katherine E. Hagerman,2010 Merrill Chemistry Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 **Basic Calculations for Chemical and Biological Analysis** Bassey J. S. Efiok,Etim Effiong Eduok,2000 Like the 1993 edition this iteration does not assume that students lab technicians and scientists have mastered

the prerequisite calculation skills for quantitative problems in the chemical biomedical sciences A new chapter focuses on using spreadsheets and laboratory information management systems Other chapters cover calculations and techniques relevant to reagents chemical reactions properties of gases and solutions pH and buffer preparation spectrophotometry enzyme assays and radioactivity Also included are derivations of some key equations quick reference guides and an index to the practical examples Efiok is with the National Heart Lung and Blood Institute National Institutes of Health Eduok is in the chemistry department at Xavier U of Louisiana c Book News Inc **Illinois Chemistry Teacher** ,1992 *The Science Teacher* ,1992 Some issues are accompanied by a CD ROM on a selected topic Holt Chemistry Ralph Thomas Myers,2004 Synthesis and Characterization of Nitric Oxide-releasing Agents/polymers for Biomedical Applications Melissa May Batchelor,2004 **Chemical Engineering Education** ,2004 *Chemical Abatement of Acid Mine Drainage Formation* Steven Jobe Onysko,1986 *Limiting Reactant and Percent Lifeliqe*,2019 This lesson plan covers analyzing chemical reactions in order to determine limiting reactants and excess reactants including calculating the amount of excess reactant calculate the theoretical yield of a reaction and calculating the percent yield of a reaction

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paragraphs ii body includes a minimum of three

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