

STOICHIOMETRY: MOLE-MOLE PROBLEMS



How many moles of hydrogen are needed to completely react with two moles of nitrogen?

$$2 \text{ mol } N_2 \left| \frac{3 \text{ mol } H_2}{1 \text{ mol } N_2} \right| = 6$$

6 moles H_2



How many moles of oxygen are produced by the decomposition of six moles of potassium chlorate?

$$6 \text{ moles } KClO_3 \times \frac{3 \text{ mol } O_2}{2 \text{ mol } KClO_3} = \frac{18}{2} = 9$$

9 moles of O_2



How many moles of hydrogen are produced from the reaction of three moles of zinc with an excess of hydrochloric acid?

$$3 \text{ mol } Zn \times \frac{1 \text{ mol } H_2}{1 \text{ mol } Zn} = 3$$

3 moles of H_2



How many moles of oxygen are necessary to react completely with four moles of propane (C_3H_8)?

$$4 \text{ mol } C_3H_8 \times \frac{5 \text{ mol } O_2}{1 \text{ mol } C_3H_8}$$

20 mol of O_2



How many moles of potassium nitrate are produced when two moles of potassium phosphate react with two moles of aluminum nitrate?

$$2 \text{ mol } K_3PO_4 \left| \frac{3 \text{ mol } KNO_3}{1 \text{ mol } K_3PO_4} \right| =$$

6 moles of KNO_3

Introduction To Stoichiometry Packet Answer Key

U. Kiencke, G. L. Gissinger



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Advances in Automotive

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web 1 ton 12 000 btu hr fig 1 illustrates some of these definitions using water as the medium experiencing a heat transfer process this graph plots the water temperature vs the enthalpy of the water heat content in btu lb we all know that water boils at 212 of atmospheric pressure at sea level

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web for r12 saturation pressure at 120 f is 172 35 psia for r134a saturation pressure at 120 f is 186 023 psia if a lower operating pressure is the only desired criterion for the design clearly r12 is a better choice c arizona s summer the design specification given thus far is the temperature of the condenser see diagram which is 120 f