

Name: _____

Date: _____

Period: _____

Atomic Structure, Valence Electrons, and the Periodic Table

Background Info:

Atoms are the smallest unit consist of three main subatomic particles: positively charged particles (protons) with neutral neutrons. Negatively charged electrons orbit the nucleus in shells. An atom's electrons determine its chemical reactivity.

Periodic Table of Elements:

Label the Periods, Families, and Valence Electrons.

1	2				
3	4				
5	6	7			
8	9	10	11	12	13
14	15	16	17	18	19
20	21	22	23	24	25
26	27	28	29	30	31
32	33	34	35	36	37

38	39	40	41	42	43
44	45	46	47	48	49

Valence Electrons:

Write the Number of Valence Electrons for each element.

S = _____ F = _____
 H = _____ Cl = _____
 Al = _____ I = _____

Reading Elements on the Periodic Table:

Fill in the Information Below

4
Bc
Beryllium
9.0122

← _____

Number of Protons (and Electrons)

← _____

Element Abbreviation

← _____

Formal Element Name

← _____

Equal to Protons + Neutrons

How to Find Protons? → Look at the Atomic Number: Atomic # _____ Protons # _____
 How to Find Electrons? → Look at the Atomic Number: Atomic # _____ Electrons # _____
 How to Find Neutrons? → Take Mass # - Atomic #: Mass # _____ Neutrons # _____

Reading Elements on the Periodic Table:

Fill in the Information Below using the Rounded Mass Number

18
Ar
Argon
39.948

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

15
P
Phosphorus
30.974

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

31
Ga
Gallium
69.723

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

53
I
Iodine
126.90

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

52
Te
Tellurium
127.60

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

55
Cs
Cesium
132.91

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

1
H
Hydrogen
1.008

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

12
Mg
Magnesium
24.305

Name: _____
 Symbol: _____
 Protons: _____
 Neutrons: _____
 Electrons: _____
 Val. e⁻: _____

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web every numeral from one through nine has touchpoints corresponding to the number s value numerals one through five have single touchpoints numerals six through nine have double touchpoints students count aloud as they touch the single touchpoints once and double touchpoints twice the numeral zero has no touchpoints and is never touched

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web mar 1 2021 so children can count without using their fingers children must count all the dots or points on the numbers for example number 1 has 1 point number 2 has 2 points number 3 has 3 points and so on meanwhile numbers 6 9 have double points so touchpoint at numbers 6 9 is counted twice

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