

Gas Laws Study Guide

Write the Law, Formula, Symbols, Units, and a sentence describing the meaning of the symbols and units.

1. A sample of gas is held at a constant volume and pressure.
2. The pressure of a gas is held constant and the volume is held constant.
3. A sample of gas is held at a constant volume and temperature.
4. A sample of gas is held at a constant pressure and temperature.
5. The volume of a gas is held constant and the pressure is held constant.
6. A sample of gas is held at a constant volume and temperature.

Name _____ Date _____

Gas Laws Study Guide

Boyle's Temperature: $T = \dots$ K $T = \dots$ $^{\circ}C$

Boyle's Pressure: $P = \dots$ atm $P = \dots$ mmHg $P = \dots$ kPa $P = \dots$ bar $P = \dots$ Pa

Boyle's Volume: $V = \dots$ L $V = \dots$ mL

Boyle's Law

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Combined Gas Law

$$P_{total} = P_1 + P_2 + P_3$$

Charles Law

$$P_1 V_1 = P_2 V_2$$

Gay-Lussac Law

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Boyle's Law

$$P_1 T_1 = P_2 T_2$$

Boyle's Law

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Boyle's Law

Complete the following statements by filling in the blanks & writing the correct units.

As Pressure increases Volume decreases (increases) given that _____ is constant.

Meaning that they have an inverse relationship.

As Volume increases Temperature increases (decreases) given that _____ is constant.

Meaning that they have a direct relationship.

As Pressure increases Temperature increases (decreases) given that _____ is constant.

Meaning that they have a direct relationship.

As the number of moles of gas in a container increases the volume increases (decreases), the Pressure increases (decreases), and the Temperature increases (decreases).

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K Morrison



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to Chemistry, Study Guide T. R. Dickson, 1999-07-07 Teaches chemistry by offering a dynamic provocative and relevant view of the topic and its importance to society and our daily lives Three themes are stressed throughout the text developing chemical thinking and chemical vision and refining problem solving skills Many chapters in this edition has been rewritten and rearranged to vitalize the topics and to include interesting examples analogies and images

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