

# Gas Laws Study Guide

Write the Law, Formula, Symbols, Units, and a sentence describing the meaning of the symbols and units.

1. A sample of gas is held at a constant volume and pressure.
2. The pressure of a gas is held constant and the volume is held constant.
3. A sample of gas is held at a constant volume and temperature.
4. A sample of gas is held at a constant pressure and temperature.
5. The volume of a gas is held constant and the pressure is held constant.
6. A sample of gas is held at a constant volume and the pressure is held constant.

Name \_\_\_\_\_ Date \_\_\_\_\_

## Gas Laws Study Guide

Boyle's Temperature:  $T = \dots$   $K$   $T = \dots$   $^{\circ}C$

Boyle's Pressure:  $P = \dots$  atm  $P = \dots$  mmHg  $P = \dots$  kPa  $P = \dots$  bar  $P = \dots$  Pa

Boyle's Volume:  $V = \dots$  L  $V = \dots$  mL

### Boyle's Law

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Combined Gas Law

$$P_{total} = P_1 + P_2 + P_3$$

Charles' Law

$$P_1 V_1 = P_2 V_2$$

Gay-Lussac's Law

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Boyle's Law

$$P_1 T_1 = P_2 T_2$$

Boyle's Law

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

Boyle's Law

Complete the following statements by filling in the blanks & writing the correct units.

As Pressure increases Volume decreases (increases) given that \_\_\_\_\_ is constant.

Meaning that they have a direct/ inverse relationship.

As Volume increases Temperature increases (decreases) given that \_\_\_\_\_ is constant.

Meaning that they have a direct/ inverse relationship.

As Pressure increases Temperature increases (decreases) given that \_\_\_\_\_ is constant.

Meaning that they have a direct/ inverse relationship.

As the number of moles of gas in a container increases the volume increases (decreases), the Pressure increases (decreases), and the Temperature increases (decreases).

# Gas Laws Study Guide

**Julie R. Frentrup**



## Gas Laws Study Guide:

The Primary FRCA Structured Oral Examination Study Guide 1 Lara Wijayasiri, Kate McCombe, Amish Patel, 2010 This fully up to date book is designed specifically for candidates preparing for the Primary FRCA structured oral examination incorporating the new exam structure and syllabus Sample questions accurately reflect the examination while model answers are systematically structured with definitions and classifications and illustrated with essential diagrams and graphs The books provide clear and concise explanations to key scientific concepts and problem based answers to clinical scenarios This first part contains questions on physiology and physics Packed with new guidelines fundamental topics that are poorly covered in other main texts and current hot topics this book and its companion The Primary FRCA Structured Oral Examination Study Guide 2 are the definitive revision aid to the Primary FRCA but will also be of value to candidates preparing for the basic science component of the Final FRCA as reference source for qualified anaesthetists and as a text for tutors preparing candidates for the structured oral examination

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## Unveiling the Magic of Words: A Review of "**Gas Laws Study Guide**"

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