

Honors Chemistry – Unit 3 Review

Complete each of the following problems before class. If you cannot complete any of them, the teacher will provide you with a copy of the problems. If you are unable to complete any of them, please bring them to class for help.

Review

Write a balanced chemical equation for the reaction of hydrogen gas with oxygen gas to form water. Write the names of the reactants and products. Balance the equation. Label the reactants and products. Write the names of the reactants and products. Write the names of the reactants and products. Write the names of the reactants and products.

1. Explain why the combustion of hydrogen gas is an exothermic reaction. (10 points)

2. Explain why the combustion of hydrogen gas is an exothermic reaction. (10 points)

3. Explain how the combustion of hydrogen gas is an exothermic reaction. (10 points)

Chemical Reaction Theory – Pressure Concepts

Write a balanced chemical equation for the reaction of hydrogen gas with oxygen gas to form water. Write the names of the reactants and products. Balance the equation. Label the reactants and products. Write the names of the reactants and products. Write the names of the reactants and products.

4. Explain why the combustion of hydrogen gas is an exothermic reaction. (10 points)

5. Explain why the combustion of hydrogen gas is an exothermic reaction. (10 points)

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Albert A Gayle



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