

## Evolution Study Guide Answer Key

1. **EVOLUTION** or change over time, is the process by which modern organisms have descended from ancient organisms
2. A scientific **THEORY** is a well supported, testable explanation of phenomena that have occurred in the natural world.
3. **CHARLES DARWIN** was an English naturalist who made numerous observations during his travels on the Beagle which led him to pose a hypothesis about how life changes over time.
4. **FOSSILS** are the preserved remains of ancient organisms that provide evidence for how life has changed over time.
5. **JEAN-BAPTISTE LAMARCK** proposed that the selective use or disuse of an organ led to a change the organ that was then passed on to the offspring.
6. **FITNESS** is the ability of an individual to survive and reproduce in its specific environment.
7. The process whereby individuals that are better suited to their environment survive and reproduce more successfully is called **NATURAL SELECTION**, which Darwin nicknamed **SURVIVAL OF THE FITTEST**.
8. Structures that have different mature forms but develop from the same embryonic tissues are called **HOMOLOGOUS STRUCTURES**.
9. Homologous structures that are so reduced in size that they no longer function, like the human appendix or legs in skinks, are called **VESTIGIAL** organs.
10. The idea that each living species has descended with changes over time from other species is called **DESCENT** with **MODIFICATION**.
11. Dolphins, penguins, and sharks are distantly-related species that share similar characteristics which help them live in water. This is an example of **CONVERGENT** evolution.
12. Even though the Galapagos finches share a common ancestor, they have evolved to fit the ecosystems of their individual islands. This is an example of **DIVERGENT** evolution.
13. Another name for divergent evolution is **ADAPTIVE RADIATION**.

### MULTIPLE CHOICE

Circle the answer that best completes the statement.

A well supported, testable explanation of phenomena that have occurred in the natural world is called a

- A. hypothesis
- B. theory**
- C. law

Of all the places he visited, the \_\_\_\_\_ Islands influenced Darwin's ideas about evolution the most.

# Evolution Study Guide Packet Answers

**Jane Holcombe, Charles Holcombe**



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 $\text{N}_2 + \text{F}_2 \rightarrow \text{NF}_3$  2  $\text{C}_6\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$  3  $\text{HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2$  4  $\text{GaBr}_3 + \text{NaBr} \rightarrow \text{SnO}_4 + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3$  5  $\text{NF}_3 + \text{SnF}_2 \rightarrow \text{N}_2\text{O}_3$  using the equation from problem 2 above answer the following

5 3 stoichiometry calculations chemistry libretexts - Mar 11 2023

web learning objectives to balance equations that describe reactions in solution to calculate the quantities of compounds produced or consumed in a chemical reaction to solve quantitative problems involving the stoichiometry of reactions in solution

*stoichiometry practice problems chemistry steps* - Jun 02 2022

web stoichiometry stoichiometry practice problems this is a comprehensive end of chapter set of practice problems on stoichiometry that covers balancing chemical equations mole ratio calculations limiting reactants and percent yield concepts the links to the corresponding topics are given below

**balancing equations and simple stoichiometry key solano** - Aug 16 2023

web balancing equations and simple stoichiometry key balance the following equations 1  $\text{N}_2 + 3\text{F}_2 \rightarrow 2\text{NF}_3$  2  $2\text{C}_6\text{H}_{10} + 17\text{O}_2 \rightarrow 12\text{CO}_2 + 10\text{H}_2\text{O}$  3  $4\text{HBr} + 1\text{KHCO}_3 \rightarrow 1\text{H}_2\text{O} + 1\text{KBr} + 1\text{CO}_2$  4  $3\text{GaBr}_3 + 3\text{Na}_2\text{SO}_3 \rightarrow 1\text{Ga}_2\text{SO}_3 + 3\text{SO}_3 + 6\text{NaBr} + 3\text{SnO}_2$  5  $3\text{NF}_3 + 3\text{SnF}_2 \rightarrow 1\text{N}_2\text{O}_3$  using the following equation  $2\text{NaOH} + \text{H}_2\text{SO}_4 \rightarrow 2\text{H}_2\text{O} + \text{Na}_2\text{SO}_4$

*ideal stoichiometry practice khan academy* - Jan 29 2022

web ideal stoichiometry google classroom you might need calculator periodic table given the following reaction  $\text{Ce} + \text{Zn} + \text{CuCl}_2 \rightarrow \text{ZnCl}_2 + \text{Cu} + \text{Zn} + \text{CuCl}_x$  2  $\text{ZnCl}_x + 2\text{Cu}$  how many moles of  $\text{Ce} + \text{ZnCl}_2 + \text{ZnCl}_x + 2\text{Cu}$  will be produced from 23.0 g of  $\text{Ce} + \text{Zn}$  assuming  $\text{Ce} + \text{CuCl}_2 + \text{CuCl}_x + 2$  is available in excess

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web jul 22 2013 balance the following equations br 1 1 n<sub>2</sub> 3 f<sub>2</sub> 2 nf<sub>3</sub> br 2 2 c<sub>6</sub>h<sub>10</sub> 17 o<sub>2</sub> 12 co<sub>2</sub> 10 h<sub>2</sub>o br 3 1 hbr 1 khco<sub>3</sub> 1 h<sub>2</sub>o 1 kbr 1 co<sub>2</sub> br 4 2 gabr<sub>3</sub> 3 na<sub>2</sub>so<sub>3</sub> 1 ga<sub>2</sub> so<sub>3</sub> 3 6 nabr br 5 3 sno 2 nf<sub>3</sub> 3 snf<sub>2</sub> 1 n<sub>2</sub>o<sub>3</sub> br using the following equation br 2 naoh h<sub>2</sub>so<sub>4</sub> 2 h<sub>2</sub>o na<sub>2</sub>so<sub>4</sub> br

**3 stoichiometry chemical formulas and equations** - May 01 2022

web the stoichiometry of a balanced chemical equation identifies the maximum amount of product that can be obtained the stoichiometry of a reaction describes the relative amounts of reactants and products in a balanced chemical equation

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web question stoichiometry practice worksheet balancing equations and simple stoichiometry balance the following equations 1 2 3 4 5 i n<sub>2</sub> 3 f<sub>2</sub> 2 nf<sub>3</sub> 2 cah 40 17 02 12 co<sub>2</sub> 10 h<sub>2</sub>o hbr khco<sub>3</sub> h<sub>2</sub>o kbr co<sub>2</sub> gabr na<sub>2</sub>so<sub>3</sub> gaz so<sub>3</sub> 3 nabr sno nf<sub>3</sub> snf<sub>2</sub>

[5 2 1 practice problems reaction stoichiometry](#) - Oct 06 2022

web may 28 2020 write the balanced equation for the reaction and calculate how much sio<sub>2</sub> is required to produce 3 00 kg of sic answer ce sio<sub>2</sub> 3c → sic 2co

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web question stoichiometry practice worksheet balancing equations and simple stoichiometry balance the following equations n<sub>2</sub> f<sub>2</sub> nf c<sub>6</sub>h<sub>10</sub> 02 c0 ho 3 hbr khco<sub>3</sub> h<sub>2</sub>o kbr 4 gabry na so<sub>3</sub> gaz so<sub>3</sub> 3 5 sno nf snf<sub>2</sub> n<sub>2</sub>o co<sub>2</sub> nabr solve the following stoichiometry grams grams problems 6

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web jun 30 2023 stoichiometry and balanced equations in stoichiometry balanced equations make it possible to compare different elements through the stoichiometric factor discussed earlier this is the mole ratio between two factors in a chemical reaction found through the ratio of stoichiometric coefficients

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web chemistry chemistry questions and answers name unit 6 balancing equations and simple stoichiometry this worksheet is worth 10 points for problems 5 7 make sure to show your work using the factor label method please circle or highlight your answers

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web gabrielle m 9 years ago i m working on chemical reactions double and single replacement on flvs now my first question for this video is how do you have 4 aluminum atoms when it says 2al subscript 2 do you just add the 2 and the subscript 2 same thing goes with the o<sub>3</sub> i am so lost

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web balancing equations and simple stoichiometry answers balance the following equations 1 1 n<sub>2</sub> 3 f<sub>2</sub> 2 nf<sub>3</sub> 2 2 c<sub>6</sub>h<sub>10</sub> 17 o<sub>2</sub>  
12 co<sub>2</sub> 10 h<sub>2</sub>o 3 1 hbr 1 khco<sub>3</sub> 1 h<sub>2</sub>o 1 kbr 1 co<sub>2</sub> 4 2 gabr<sub>3</sub> 3 na<sub>2</sub>so<sub>3</sub> 1 ga<sub>2</sub> so<sub>3</sub> 3 6 nabr 5 3 sno 2 nf<sub>3</sub> 3 snf<sub>2</sub> 1 n<sub>2</sub>o<sub>3</sub>

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web google classroom balance the following chemical equation text mg oh 2 mg oh 2 text hcl rightarrow hcl text mgcl 2  
mgcl<sub>2</sub> text h 2 text o h<sub>2</sub>o note all reactants and products require a coefficient of at least one stuck review related articles  
videos or use a hint report a problem 7 4 1 x x y y theta  $\theta$  pi  $\pi$  8 5 2 0 9 6

stoichiometry article chemical reactions khan academy - Jul 15 2023

web a common type of stoichiometric relationship is the mole ratio which relates the amounts in moles of any two substances  
in a chemical reaction we can write a mole ratio for a pair of substances by looking at the coefficients in front of each species  
in

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molecular composition unit 4 mass spectrometry unit 5 chemical reactions and stoichiometry unit 6 more about chemical  
reactions unit 7 electronic structure of atoms unit 8 periodic table