

QUESTION – CHEMISTRY

FORM -2

CHEMICAL FAMILIES

1. Study the information in the table below and answer the questions that follow:

Element	Atomic radius (nm)	Ionic radius (nm)
W	0.114	0.195
X	0.072	0.136
Y	0.133	0.216
Z	0.099	0.181

(a) Would these form part of a metallic or a non-metallic group? Explain

(b) Suggest an element in the table above likely to be the most reactive. Explain

2. State the reason for using Argon in electric light bulbs

3. Study the information in the table below and answer the questions that follow. The letters do not represent the actual symbols of the elements.

Element	Electronic configuration	Boiling point
X	2.7	-188°C
Y	2.8.7	-35°C
Z	2.8.8.7	59°C

(a) What is the general name given to the group in which the elements X, Y and Z belong?

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