

# Graphing Periodic Trends Worksheet

1. Using the ionization energy (IE) values of the elements in Period 2 listed below, make a line graph of IE vs. the atomic number.

Period 2					
Element	Atomic Number	IE (kJ/mol)	Element	Atomic Number	IE (kJ/mol)
Li	3	519	Li	3	519
Be	4	900	Be	4	900
B	5	799	B	5	799
C	6	1088	C	6	1088



- a) What trend do you notice for the ionization energies in Period 2?
- b) Explain why you observe this trend.
2. Make a scatter plot of ionization energy vs. atomic number for the first three atoms in Group 2A listed below.

Group 2A		
Element	Atomic Number	IE (kJ/mol)
Be	4	900
Mg	12	736
Ca	20	590



# Graphing Activities Chemistry Answer Key

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## **Table of Contents Graphing Activities Chemistry Answer Key**

1. Understanding the eBook Graphing Activities Chemistry Answer Key
  - The Rise of Digital Reading Graphing Activities Chemistry Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Graphing Activities Chemistry Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Graphing Activities Chemistry Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Graphing Activities Chemistry Answer Key
  - Personalized Recommendations
  - Graphing Activities Chemistry Answer Key User Reviews and Ratings
  - Graphing Activities Chemistry Answer Key and Bestseller Lists
5. Accessing Graphing Activities Chemistry Answer Key Free and Paid eBooks
  - Graphing Activities Chemistry Answer Key Public Domain eBooks
  - Graphing Activities Chemistry Answer Key eBook Subscription Services
  - Graphing Activities Chemistry Answer Key Budget-Friendly Options

6. Navigating Graphing Activities Chemistry Answer Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Graphing Activities Chemistry Answer Key Compatibility with Devices
  - Graphing Activities Chemistry Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Graphing Activities Chemistry Answer Key
  - Highlighting and Note-Taking Graphing Activities Chemistry Answer Key
  - Interactive Elements Graphing Activities Chemistry Answer Key
8. Staying Engaged with Graphing Activities Chemistry Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Graphing Activities Chemistry Answer Key
9. Balancing eBooks and Physical Books Graphing Activities Chemistry Answer Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Graphing Activities Chemistry Answer Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Graphing Activities Chemistry Answer Key
  - Setting Reading Goals Graphing Activities Chemistry Answer Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Graphing Activities Chemistry Answer Key
  - Fact-Checking eBook Content of Graphing Activities Chemistry Answer Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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