

Skills Worksheet

Concept Review Chapter 5: Section 1

Section: Simple Ions

Complete each statement below by choosing a term from the following list. Terms may be used more than once.

- | | | | |
|----|----|-------|--------|
| 10 | 11 | ion | octet |
| 12 | 13 | anion | cation |

1. An atom or group of atoms that has a positive or negative electric charge because it has lost or gained electrons is a(n) ion.
2. The sodium ion has 11 protons, 12 neutrons, and 10 electrons.
3. In most chemical reactions, atoms tend to match the outer electron configuration of the noble gases. This is called the octet rule.

Complete each statement below by writing the correct term or phrase.

4. An anion is an ion with a negative charge.
5. Valence electrons occupy the outermost energy level of an atom.
6. The chloride ion and the chlorine atom have different electron configurations.
7. A cation is an ion with a positive charge.
8. Many stable ions have an electron configuration of a noble gas.

Answer the following questions in the space provided.

9. How do the outer-shell electron configurations for ions of Group 1, Group 2, and Group 15, Group 16, and Group 17 elements compare with those of the noble gases?
They are the same. Group 1 + 2 ions have noble gas configuration of previous row. Group 15, 16 + 17 ions have noble gas configuration in their row.

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Table of Contents Holt Chemistry Answers Pg 445

1. Understanding the eBook Holt Chemistry Answers Pg 445
 - The Rise of Digital Reading Holt Chemistry Answers Pg 445
 - Advantages of eBooks Over Traditional Books
2. Identifying Holt Chemistry Answers Pg 445
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Holt Chemistry Answers Pg 445
 - User-Friendly Interface
4. Exploring eBook Recommendations from Holt Chemistry Answers Pg 445
 - Personalized Recommendations
 - Holt Chemistry Answers Pg 445 User Reviews and Ratings

- Holt Chemistry Answers Pg 445 and Bestseller Lists
- 5. Accessing Holt Chemistry Answers Pg 445 Free and Paid eBooks
 - Holt Chemistry Answers Pg 445 Public Domain eBooks
 - Holt Chemistry Answers Pg 445 eBook Subscription Services
 - Holt Chemistry Answers Pg 445 Budget-Friendly Options
- 6. Navigating Holt Chemistry Answers Pg 445 eBook Formats
 - ePub, PDF, MOBI, and More
 - Holt Chemistry Answers Pg 445 Compatibility with Devices
 - Holt Chemistry Answers Pg 445 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Holt Chemistry Answers Pg 445
 - Highlighting and Note-Taking Holt Chemistry Answers Pg 445
 - Interactive Elements Holt Chemistry Answers Pg 445
- 8. Staying Engaged with Holt Chemistry Answers Pg 445
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Holt Chemistry Answers Pg 445
- 9. Balancing eBooks and Physical Books Holt Chemistry Answers Pg 445
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Holt Chemistry Answers Pg 445
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Holt Chemistry Answers Pg 445
 - Setting Reading Goals Holt Chemistry Answers Pg 445
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Holt Chemistry Answers Pg 445
 - Fact-Checking eBook Content of Holt Chemistry Answers Pg 445
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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