

# CHEMISTRY MULTIPLE CHOICE QUESTIONS

## A. Atoms, molecules and stoichiometry

2002 -2019



C

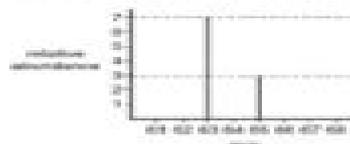
As a consequence, an adult human can be considered to have a daily diet of 1.80 kg of carbohydrate (empirical formula  $C_6H_{12}O_6$ ).

Which mass of carbon dioxide does a person produce each day if all the carbohydrate eaten is digested and oxidised?

- A 0.907 kg    B 0.906 kg    C 1.70 kg    D 3.54 kg

(2002 NA/12)

The diagram shows the mass spectrum of a sample of naturally-occurring copper.



What is the relative atomic mass of this copper?

- A 63.2    B 63.5    C 63.8    D 64.0

(2002 NA/12)

Which pairs of compounds have the same empirical formula?

- ethane and ethene
- ethane and cyclohexane
- cyclohexane and oct-1-ene

(2002 NA/12)

A stream of  $10\text{ cm}^3$  of methane and  $10\text{ cm}^3$  of ethane was sparked with an excess of oxygen. After cooling to room temperature, the residual gas was passed through aqueous potassium hydroxide.

Which compounds have the empirical formula  $CH_2O$ ?

- methanol
- ethanoic acid
- methyl methanoate

(2002 ON/11)

What is the number of molecules in  $500\text{ cm}^3$  of oxygen under room conditions?

- A  $1.25 \times 10^{23}$   
 B  $1.25 \times 10^{24}$   
 C  $3.0 \times 10^{23}$   
 D  $3.0 \times 10^{24}$

(2004 MO/12)

In the preparation of soft margarine, glycerol trioleate



is fully hydrogenated so that, on average, one of its side-chains is converted into the  $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CO}_2\text{C}_2\text{H}_5$  moiety and two side-chains are converted into the  $\text{CH}_3(\text{CH}_2)_7\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{CO}_2\text{C}_2\text{H}_5$  moiety.

How many moles of hydrogen are required to convert one mole of glycerol trioleate into the soft margarine?

- A 4    B 5    C 6    D 8

(2008 MO/12)

Use of the Data Booklet is relevant to this question.

Analysed classes may obtain more useful accounts of extra work. Access to a 100% and

# Multiple Choice Question For Instrumental Chemistry

**M Tight**



**Multiple Choice Question For Instrumental Chemistry:**

## Whispering the Strategies of Language: An Emotional Quest through **Multiple Choice Question For Instrumental Chemistry**

In a digitally-driven world where displays reign great and quick interaction drowns out the subtleties of language, the profound secrets and emotional subtleties hidden within words usually get unheard. However, located within the pages of **Multiple Choice Question For Instrumental Chemistry** a charming fictional treasure blinking with raw thoughts, lies an exceptional journey waiting to be undertaken. Composed by a talented wordsmith, this enchanting opus encourages visitors on an introspective journey, gently unraveling the veiled truths and profound affect resonating within the very cloth of each word. Within the mental depths of the poignant evaluation, we will embark upon a honest exploration of the book is primary styles, dissect their captivating publishing model, and yield to the strong resonance it evokes deep within the recesses of readers hearts.

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