

# CLASSIFYING MATTER

## READING COMPREHENSION WORKSHEET

<p>Complete each sentence by filling in the missing information.</p> <p>1. A pure substance made of only one type of atom is called _____.</p> <p>2. How is matter classified? _____</p> <p>3. The number of protons in the nucleus of an atom is called its _____.</p> <p>4. A neutron has the same mass as a proton but no _____.</p> <p>5. The number of electrons in an atom is equal to the number of _____.</p> <p>Draw a line from each element to its correct family.</p> <p>1. Helium 2. Sodium 3. Chlorine 4. Argon 5. Potassium 6. Bromine</p> <p>Answer in 20 complete sentences.</p> <p>1. Why is hydrogen a different element than helium?</p> <p>2. How does chlorine differ from sodium in its bonding?</p> <p>3. How is a carbon atom different from a sodium atom?</p> <p>4. How does the number of protons in an atom affect its chemical properties?</p>	<p>Read each statement and decide if it is True or False.</p> <p>1. Elements are made of only one kind of atom.  <input type="checkbox"/> True  <input type="checkbox"/> False</p> <p>2. Atoms are the smallest particles of matter.  <input type="checkbox"/> True  <input type="checkbox"/> False</p> <p>3. Protons and neutrons have equal masses.  <input type="checkbox"/> True  <input type="checkbox"/> False</p> <p>4. Electrons are the smallest particles of matter.  <input type="checkbox"/> True  <input type="checkbox"/> False</p> <p>5. Helium is an inert gas.  <input type="checkbox"/> True  <input type="checkbox"/> False</p> <p>6. Chlorine is a halogen.  <input type="checkbox"/> True  <input type="checkbox"/> False</p> <p>7. Argon is a noble gas.  <input type="checkbox"/> True  <input type="checkbox"/> False</p> <p>8. Sodium is a metal.  <input type="checkbox"/> True  <input type="checkbox"/> False</p> <p>9. Potassium is a metal.  <input type="checkbox"/> True  <input type="checkbox"/> False</p> <p>10. Bromine is a non-metal.  <input type="checkbox"/> True  <input type="checkbox"/> False</p>	<p>1. Elements</p> <p>2. Atoms</p> <p>3. Molecules</p> <p>4. Compounds</p> <p>5. Mixtures</p> <p>6. Pure substances</p> <p>7. Elements</p> <p>8. Compounds</p> <p>9. Mixtures</p> <p>10. Pure substances</p> <p>11. Elements</p> <p>12. Compounds</p> <p>13. Mixtures</p> <p>14. Pure substances</p> <p>15. Elements</p> <p>16. Compounds</p> <p>17. Mixtures</p> <p>18. Pure substances</p> <p>19. Elements</p> <p>20. Compounds</p> <p>21. Mixtures</p> <p>22. Pure substances</p>	<h3>Classifying Matter</h3> <p>Read the passage carefully before answering the questions that follow.</p> <div style="border: 1px solid black; padding: 5px;"> <p><b>Working Passage</b></p> <p>Matter is anything that has mass and occupies space. It is made up of particles. Atoms and molecules are the smallest particles of matter that cannot be created or destroyed. They are the building blocks of matter. Matter is classified into pure substances and mixtures. Pure substances are made of only one kind of particle. They have a fixed composition and definite physical and chemical properties. They are further classified into elements and compounds. Elements are made of only one kind of atom and cannot be broken down further by chemical means, such as heating or cooling. Compounds are made of two or more elements chemically combined together. They can be broken down into simpler substances by chemical means. Matter is further classified into solids, liquids, and gases. Solids have a definite shape and volume. Liquids have a definite volume but no definite shape. Gases have neither a definite shape nor a definite volume. Matter is further classified into pure substances and mixtures. Pure substances are made of only one kind of particle. They have a fixed composition and definite physical and chemical properties. They are further classified into elements and compounds. 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They can be broken down into simpler substances by chemical means.</p> </div> <p>Answer the questions that have completed the statement.</p> <p>1. Matter is anything that has mass and occupies space.</p> <p>2. Matter is classified into pure substances and mixtures.</p> <p>3. Pure substances are made of only one kind of particle.</p> <p>4. Pure substances have a fixed composition and definite physical and chemical properties.</p> <p>5. Pure substances are further classified into elements and compounds.</p> <p>6. Elements are made of only one kind of atom and cannot be broken down further by chemical means, such as heating or cooling.</p> <p>7. Compounds are made of two or more elements chemically combined together.</p> <p>8. Compounds can be broken down into simpler substances by chemical means.</p> <p>9. Matter is further classified into solids, liquids, and gases.</p> <p>10. Solids have a definite shape and volume.</p> <p>11. Liquids have a definite volume but no definite shape.</p> <p>12. Gases have neither a definite shape nor a definite volume.</p> <p>13. Matter is further classified into pure substances and mixtures.</p> <p>14. Pure substances are made of only one kind of particle.</p> <p>15. Pure substances have a fixed composition and definite physical and chemical properties.</p> <p>16. Pure substances are further classified into elements and compounds.</p> <p>17. Elements are made of only one kind of atom and cannot be broken down further by chemical means, such as heating or cooling.</p> <p>18. Compounds are made of two or more elements chemically combined together.</p> <p>19. Compounds can be broken down into simpler substances by chemical means.</p> <p>20. Matter is further classified into solids, liquids, and gases.</p> <p>21. Solids have a definite shape and volume.</p> <p>22. Liquids have a definite volume but no definite shape.</p> <p>23. Gases have neither a definite shape nor a definite volume.</p> <p>24. Matter is further classified into pure substances and mixtures.</p> <p>25. Pure substances are made of only one kind of particle.</p> <p>26. Pure substances have a fixed composition and definite physical and chemical properties.</p> <p>27. Pure substances are further classified into elements and compounds.</p> <p>28. Elements are made of only one kind of atom and cannot be broken down further by chemical means, such as heating or cooling.</p> <p>29. Compounds are made of two or more elements chemically combined together.</p> <p>30. Compounds can be broken down into simpler substances by chemical means.</p>
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Holt Science Spectrum Classifying Matter Answer Key: Bestsellers in 2023 The year 2023 has witnessed a remarkable surge in literary brilliance, with numerous engrossing novels captivating the hearts of readers worldwide. Lets delve into the realm of top-selling books, exploring the fascinating narratives that have enthralled audiences this year. Holt Science Spectrum Classifying Matter Answer Key : Colleen Hoover's "It Ends with Us" This poignant tale of love, loss, and resilience has gripped readers with its raw and emotional exploration of domestic abuse. Hoover skillfully weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can triumph. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This captivating historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids captivating storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Holt Science Spectrum Classifying Matter Answer Key : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, entrancing readers with its evocative prose and mesmerizing setting. These top-selling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of captivating stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and thrilling novel that will keep you wondering until the very end. The novel is a cautionary tale about the dangers of obsession and the power of evil.

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