

General Chemistry II Jasperse  
Acid-Base Chemistry: Extra Practice Problems

General Types/Groups of problems:

Conceptual Questions: Acids, Bases, and Conjugates, Miscellaneous	p1	$K_b$ and $pK_b$ , Base Strength, and using $K_b$ or $pK_b$ to Calculate $[OH^-]$ , $pOH$ , $pH$ , and/or $[H^+]$	p7-10
Recognizing Strong versus Weak Acids; Recognizing Basic versus Nonbasic	p3	Recognizing Acid/Base Properties when Ionics are Dissolved in Water	p11
pH Calculations: Relationships between pH and pOH	p4	Answers	p12
$K_a$ , Sense + Calculations: Using $K_a$ or $pK_a$ to Calculate $[H^+]$ and/or pH; using pH to calculate $K_a$ or $pK_a$	p5		

**Conceptual Questions: Acids, Bases, and Conjugates, Miscellaneous**

- In the Brønsted-Lowry definition of acids and bases, an acid \_\_\_\_\_
  - is a proton donor.
  - is a proton acceptor.
  - forms stable hydrogen bonds.
  - breaks stable hydrogen bonds.
  - corrodes metals.
- In the Brønsted-Lowry definition of acids and bases, a base \_\_\_\_\_
  - is a proton donor.
  - is a proton acceptor.
  - forms stable hydrogen bonds.
  - breaks stable hydrogen bonds.
  - corrodes metals.
- In the following reaction in aqueous solution, the acid reactant is \_\_\_\_\_ and its conjugate base product is \_\_\_\_\_



- $CH_3COOH$ ;  $CH_3COO^-$
  - $CH_3COOH$ ;  $NH_4^+$
  - $NH_3$ ;  $CH_3COO^-$
  - $NH_3$ ;  $NH_4^+$
  - $CH_3COOH$ ;  $H_3O^+$
- In the following reaction in aqueous solution, the acid reactant is \_\_\_\_\_, and its conjugate base product is \_\_\_\_\_



- $CH_3NH_2$ ;  $CH_3NH_3^+$
  - $CH_3NH_2$ ;  $SO_4^{2-}$
  - $HSO_4^-$ ;  $CH_3NH_3^+$
  - $HSO_4^-$ ;  $SO_4^{2-}$
  - $HSO_4^-$ ;  $H_3O^+$
- Which of the following is the conjugate acid of the hydrogen phosphate ion,  $HPO_4^{2-}$ ?
    - $H_2PO_4$
    - $H_2PO_4^-$
    - $HPO_4^{2-}$
    - $PO_4^{3-}$
    - $H_3O^+$

# General Chemistry Acid Base Practice Problems

**Zhe Xu**



**General Chemistry Acid Base Practice Problems:**

## Unveiling the Power of Verbal Artistry: An Emotional Sojourn through **General Chemistry Acid Base Practice Problems**

In a global inundated with screens and the cacophony of quick communication, the profound energy and emotional resonance of verbal artistry frequently diminish into obscurity, eclipsed by the regular assault of noise and distractions. Yet, situated within the lyrical pages of **General Chemistry Acid Base Practice Problems**, a fascinating work of fictional elegance that impulses with raw emotions, lies an unique journey waiting to be embarked upon. Composed by way of a virtuoso wordsmith, this exciting opus manuals visitors on a mental odyssey, delicately revealing the latent possible and profound impact embedded within the delicate web of language. Within the heart-wrenching expanse with this evocative analysis, we will embark upon an introspective exploration of the book is main styles, dissect their interesting writing style, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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