

Laboratory Exercise: Double Replacement Reactions...

TABLE II: SCALE UP OF PRECIPITATION REACTIONS

insoluble soluble

Run #	Chemical Formula of Reagent	Chemical Formula of Reagent	List the four ions present in reagents		Formula of Solid Product appearance in test tube	Formula of Aqueous Product appearance in test tube
2	NaOH	FeCl ₃	Na ⁺	OH ⁻	Fe(OH) ₃ very small orange	very cloudy orange
			Fe ⁺³	Cl ⁻		
3	NaNO ₃	FeCl ₃	Na ⁺	NO ₃ ⁻	solid white powder	yellow cloudy
			Fe ⁺³	Cl ⁻		
5	Na ₂ CO ₃	FeCl ₃	Na ⁺	CO ₃ ⁻²	solid orange powdery	very cloudy orange
			Fe ⁺³	Cl ⁻¹		
6	Na ₂ PO ₄	FeCl ₃	Na ⁺	PO ₄ ⁻³	yellowish white small	yellow cloudy
			Fe ⁺³	Cl ⁻¹		
12	AgNO ₃	NaOH			very small brown	cloudy brown
13	NaOH	CuSO ₄	N		gel like chunk	blue uncent
17	AgNO ₃	Na ₂ CO ₃			white flake small	white cloudy
18	AgNO ₃	Na ₂ PO ₄			yellow egg	clear yellow
19	CuSO ₄	Na ₂ CO ₃			small blue/white powder	cloudy blue
20	CuSO ₄	Na ₃ PO ₄			very small blue particles	cloudy light blue

Lab 27 Double Replacement Reaction Answers

WJ Hussar



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