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labflow



Figure 4 Boiling water bath setup

Objectives

You will prepare a calibration curve for your thermometer using liquid. Before preparation of the data, the value of the boiling point must be adjusted for atmospheric pressure. Although both melting points and boiling points change with varying pressure, the dependence of the melting point on pressure is very small and negligible. However, the rate of change of the boiling point with pressure is large. Atmospheric pressure at sea level is about 1013 mm. The boiling point of water at 1013 mm is equal to 100°C. However, the lab is near sea level so atmospheric pressure will be slightly lower than 1013 mm. During the graph, you will fit the atmospheric pressure to get the boiling point of water at your location. The atmospheric pressure can vary slightly during the process. Please remember to record the atmospheric pressure on your report sheet during pre-lab (20), we will use the average of 20(17) for each run even that atmospheric pressure is below 1013 mm. The

$$P_{\text{atm}} = (760 \text{ mm}) - (\text{atmospheric pressure}) = 0.627 \frac{\text{mm}}{\text{mm}} \quad (1.0)$$

is subtracted from 100°C to get the boiling point of water at the process.

For example, if atmospheric pressure was 120 mm, the correction for

$$P_{\text{atm}} = (760 \text{ mm}) - (120 \text{ mm}) = 0.627 \frac{\text{mm}}{\text{mm}} = 0.507$$

The boiling point at 120 mm would be $100.00^\circ\text{C} - 0.507^\circ\text{C} = 99.49^\circ\text{C}$.

20. Change the axes titles in the spreadsheet to the appropriate units. Change the units and explore the text with "Precision" on the horizontal axis and on the vertical axis.

Click on the side of the graph and change it to your name. Please will be the name of your graph if the side is not your name!



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