

Assessment

Chapter Test A

Chapter: Acids and Bases

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

- _____ 1. Which of the properties listed below is *not* characteristic of an acid?
- a sour taste
 - a slippery feel
 - the ability to conduct an electric current
 - reactivity with metals
- _____ 2. Which is an example of a weak acid?
- HCl
 - H_2SO_4
 - H_3PO_4
 - HNO_3
- _____ 3. Acetic acid is found in
- vinegar.
 - the stomach.
 - antacids.
 - oranges.
- _____ 4. A characteristic of an Arrhenius base is that it
- is an electrolyte.
 - tastes bitter.
 - dissociates to form OH^- ions.
 - All of the above
- _____ 5. Which of the following is a Brønsted-Lowry base?
- HCl
 - HCO_3^-
 - H_3O^+
 - H_3PO_4
- _____ 6. In the reaction represented by the equation
- $$\text{H}_2\text{C}_2\text{O}_4(\text{aq}) + \text{CH}_3\text{NH}_2(\text{aq}) \rightleftharpoons \text{HC}_2\text{O}_4^-(\text{aq}) + \text{CH}_3\text{NH}_3^+(\text{aq}),$$
- which of these is a conjugate acid-base pair?
- $\text{H}_2\text{C}_2\text{O}_4(\text{aq})$ and $\text{CH}_3\text{NH}_2(\text{aq})$
 - $\text{H}_2\text{C}_2\text{O}_4(\text{aq})$ and $\text{CH}_3\text{NH}_3^+(\text{aq})$
 - $\text{CH}_3\text{NH}_2(\text{aq})$ and $\text{CH}_3\text{NH}_3^+(\text{aq})$
 - $\text{HC}_2\text{O}_4^-(\text{aq})$ and $\text{CH}_3\text{NH}_3^+(\text{aq})$

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D Keegan



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