

Mole Calculations Review

Molar Mass Practice

Find the molar masses of the following compounds:

- | | | | |
|---|--|--|--------------------------------------|
| 1) NaBr | 2) PbSO ₄ | 3) Ca(OH) ₂ | 4) Na ₃ PO ₄ |
| 5) (NH ₄) ₂ CO ₃ | 6) C ₆ H ₁₂ O ₆ | 7) Fe ₂ (PO ₄) ₂ | 8) (NH ₄) ₂ S |
| 9) Zn(C ₂ H ₃ O ₂) ₂ | 10) AgF | | |

Answers (rounded to the nearest 0.1 grams)

- 1) 102.9 g/mol, 2) 303.3 g/mol, 3) 74.1 g/mol, 4) 164.0 g/mol
5) 96.0 g/mol, 6) 180.0 g/mol, 7) 357.4 g/mol, 8) 68.1 g/mol
9) 183.4 g/mol, 10) 126.9 g/mol

Mass - Moles - Particles Practice

- How many moles of Na are in 42 g of Na?
- How many moles of O are in 8.25 g of O?
- What is the mass of 2.18 mol of Cu?
- What is the mass of 0.28 mol of iron?
- How many atoms are in 7.2 mol of chlorine?
- How many atoms are in 36 g of bromine?
- How many moles are in 1.0×10^9 atoms?
- What is the mass of 1.20×10^{25} atoms of sulfur?
- How many moles of CO molecules are in 52 g of CO?
- How many moles of C₂H₆ are in 124 g?
- How many moles of CCl₄ are there in 56 g of CCl₄?
- What is the mass of 2.50 mol of H₂SO₄?
- What is the mass of 0.25 mol of Fe₂O₃?
- How many molecules are there in 52 g of CO?
- How many formula units are in 22.4 g SnO₂?
- How many molecules are in 116 g CCl₄?
- What is the mass of 3.01×10^{23} formula units of Fe₂O₃?
- What is the mass of 1.2×10^{25} molecules of CO?
- How many O atoms are in 1.25 mol of SO₂?
- How many moles of O atoms are there in 1.20×10^{25} N₂O₅ molecules?
- How many formula units are in 5.33 mol of CuCl₂?
- How many copper atoms are in 5.33 mol of CuCl₂?
- How many moles of Cl atoms are in 5.33 mol of CuCl₂?

Molar Mass Practice With Answers

O García



Molar Mass Practice With Answers:

This book delves into Molar Mass Practice With Answers. Molar Mass Practice With Answers is an essential topic that needs to be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Molar Mass Practice With Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to Molar Mass Practice With Answers
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 - Chapter 5: Conclusion
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