

## The Mole Test 3

**Instructions:** There have to identical units in pairs, then make sure all the moles are balanced in ratios. Molar mass can be used to convert grams to moles or moles to grams. Molar volume can be used to convert moles to litres or litres to moles. Avogadro's number can be used to convert moles to particles. Atoms can be used to convert moles to particles. If you're not sure what to do, then use the molar mass.

Place in Solution in Particles

1) How many particles are in 2.00 moles of Ag<sub>2</sub>SO<sub>4</sub>?

2.00 mol	2.00 × 10 <sup>23</sup> particles	2.00 × 10 <sup>24</sup> particles
	2.00 × 10 <sup>23</sup> particles	2.00 × 10 <sup>24</sup> particles

2) How many moles are 1.00 × 10<sup>23</sup> particles of CO<sub>2</sub>?

1.00 × 10 <sup>23</sup> particles	1.00 mol	1.00 × 10 <sup>23</sup> mol
	1.00 × 10 <sup>23</sup> particles	1.00 × 10 <sup>23</sup> mol

3) How many particles are in 2.00 moles of Ag<sub>2</sub>SO<sub>4</sub>?

2.00 mol	2.00 × 10 <sup>23</sup> particles	2.00 × 10 <sup>24</sup> particles
	2.00 mol	2.00 × 10 <sup>24</sup> particles

4) How many moles are 1.00 × 10<sup>23</sup> particles of CO<sub>2</sub>?

1.00 × 10 <sup>23</sup> particles	1.00 mol	1.00 × 10 <sup>23</sup> mol
	1.00 × 10 <sup>23</sup> particles	1.00 × 10 <sup>23</sup> mol

Place in Solution in Mass:

Calculate the molar masses for the following substances. Show your work!

<u>Chemical Formula</u>	<u>Molar Mass (g/mol)</u>
1) Ag <sub>2</sub> SO <sub>4</sub>	Ag = 107.87 g/mol S = 32.06 g/mol O = 16.00 g/mol Ag <sub>2</sub> SO <sub>4</sub> = 2(107.87) + 32.06 + 4(16.00) = 311.80 g/mol

# Mole Test Answer

**J Rink**



**Mole Test Answer:**

## **Mole Test Answer** Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the ability of words has be evident than ever. They have the capacity to inspire, provoke, and ignite change. Such could be the essence of the book **Mole Test Answer**, a literary masterpiece that delves deep into the significance of words and their effect on our lives. Compiled by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we will explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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