

Lab 8, MOLE RATIOS IN A CHEMICAL REACTION



MATERIALS

150 mL beaker, pipet, small beaker, balance, weighing paper, hotplate, sodiumhydrogencarbonate (s) and HCl (aq) 3.0 M.

SAFETY

STUDENTS MUST WEAR GOGGLES, APRONS AND GLOVES THROUGHOUT, DON NOT REMOVE UNTILL YOU ARE TOLD TO DO SO.

PROCEDURE

1. Mass a 150.0 mL beaker, record the mass in your lab notebook.
2. Mass 2.00 grams of NaHCO_3 in the beaker.
3. Obtain about 30. mL of 3.0 M HCl in a small beaker. Keep the beaker in your sink when not in use.
4. Obtain a plastic pipet, fill it with HCl solution.

Mole Ratios Packet Model A Chemical Reaction

Lingjun Ying



Mole Ratios Packet Model A Chemical Reaction:

Mole Ratios Packet Model A Chemical Reaction Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is actually remarkable. This extraordinary book, aptly titled "**Mole Ratios Packet Model A Chemical Reaction**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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