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web mar 27 2008 table 2 determining the atomic mass of beanium isotopes and molar mass of beanium bean type number of beans mass of beans of this kind atomic mass of this bean isotope abundanc e kidney bean 32 36 1 122 black eye bean 98 23 0 25 white bean 30 16 0 50 calculation of atomic mass of beanium 36 32 1 amu

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web calculate theaverageatomic massofbeaniumtobeplaced ontheperiodic table beanium isotopelabclass set data totalmass atoms total mass isotope 1 atoms 1 avg mass 1 total mass isotope 2 atoms 2 avg mass 2 total mass isotope 3 atoms 3 avg mass 3 total mass ofallatoms total of atoms

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web isotopes and weighted average atomic mass amu atomic mass unit mass of one proton mass of one neutron a sample of cesium cs has the following abundance cs 132 20 0 cs 133 75 3 cs 134 4 7 weighted average atomic mass $132 \times 0.200 + 133 \times 0.752 + 134 \times 0.047 = 132.85$ amu

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web beanium isotope lab introduction purpose what is an isotope what does it mean to say that the atoms in a sample of an element are isotopes of each other ordinary beans are a lot bigger than atoms but perhaps they can give you one or two clues about isotopes we have talked in class about isotopes atoms of the same element that differ in

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web beanium isotope lab section 1 pre lab questions define average atomic mass write a mathematical equation that shows how you would determine the average atomic mass of an element section 2 data table don't forget calibration and units total all beans 100 beanium isotope blackium beanium isotope brownium beanium

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web beanium isotope b brownium beanium isotope c whiteium of atoms beans mass beans in cup mass empty cup mass of atoms beans 2 minus 3 average mass of 1 bean 4 divided by 1 of beans 1 a b c $\times 100$ section 3 calculate the average atomic mass of beanium

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web beanium isotope laba introduction purpose this lab exercise is designed to show you isotopes of an element in a simulation form you will be asked to gather data about the isotopes and organize the data if atoms were as large as beans they could be sorted counted and massed

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web in the beanium lab students obtained a sample of beanium a fictitious element in order to learn about isotopes and average atomic mass beanium has three isotopes white brown and speckled which have different masses and different percent abundances

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web your final assignment is to calculate the average atomic mass of beanium to be placed on the periodic table the formula is abundance of isotope 1 \times mass of isotope 1 abundance of isotope 2 \times mass of isotope 2 abundance of isotope 3 \times mass of

isotope 3 average isotope mass

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web 1 obtain a sample of beanium separate the pintonium navium and lentilium and then mass each group record the values in the data table 2 count the number of each isotope by counting the pintonium navium and lentilium 3 calculate the total number of isotopes

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web procedure obtain a sample of the classroom mixture of beanium sort your beanium into its three isotopic bean types count the number of beans in each pile find the mass of each pile of beans determine the average mass of each type of bean based on the samples masses

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web terms in this set 14 what are the objective of the lab calculate from experimental data the atomic mass of the fictitious element beanium by using a weighted average and simple average decide which method gives a better representation of the atomic mass of beanium

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web good luck saftey 1 do not eat the beans 2 do not throw the beans 3 do not crush the beans don t mess with the beans pre
lab questions 1 there are many beans in your container each bean represents an 2 there are three kinds of beans in your
container each kind represents a different