

# F321: Atoms, Bonds and Groups

## Atoms, Isotopes and Relative Atomic Masses

70 Marks

1. Isotopes of europium have differences and similarities:

- (i) In terms of protons, neutrons and electrons, how is an atom of  $^{151}\text{Eu}$  **different** from an atom of  $^{153}\text{Eu}$ ?

.....  
.....

[1]

- (ii) In terms of protons, neutrons and electrons, how is an atom of  $^{151}\text{Eu}$  **similar** to an atom of  $^{153}\text{Eu}$ ?

.....  
.....

[1]

[Total 2 marks]

2. Europium, atomic number 63, is used in some television screens to highlight colours. A chemist analysed a sample of europium using mass spectrometry. The results are shown in the table below.

isotope	relative isotopic mass	abundance (%)
$^{151}\text{Eu}$	151.0	47.77
$^{153}\text{Eu}$	153.0	52.23

- (a) Define the term *relative isotopic mass*.

.....  
.....  
.....  
.....

[2]

# F321 June 2014 Qp

**Lingjun Ying**



**F321 June 2014 Qp:**

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In a world inundated with displays and the cacophony of instantaneous connection, the profound energy and emotional resonance of verbal art frequently disappear into obscurity, eclipsed by the constant onslaught of sound and distractions. However, set within the lyrical pages of **F321 June 2014 Qp**, a interesting work of fictional beauty that pulses with organic emotions, lies an unique trip waiting to be embarked upon. Written by way of a virtuoso wordsmith, this exciting opus guides readers on a psychological odyssey, delicately revealing the latent potential and profound influence embedded within the intricate internet of language. Within the heart-wrenching expanse with this evocative analysis, we shall embark upon an introspective exploration of the book is key subjects, dissect its interesting publishing fashion, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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